



A Text-Book of Experimental Chemistry; (With Descriptive Notes) for Students of General Inorganic Chemistry

By Edwin Lee

Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1908 edition. Excerpt: . (nitrogen peroxide), N₂O, and N₂O₃. Heat about 8 grams of dry lead nitrate in a test tube. Note the color and odor of the gas evolved. The vapor may be condensed in a U-tube surrounded by a freezing mixture, or a portion of it may be conducted into a concentrated solution of NaOH contained in a test tube. If all of the gas is not absorbed by the solution, test the escaping gas for oxygen. Results? What is the name of the gas? Examine the substance which remains in the ignition tube. What is it? What is the name and formula of the evolved gas? Write equations for all reactions. Keep the NaOH solution for Exp. XVII. Label-it. (N₂O₂) =--8,125 cal. (e) Nitrogen pentoxide (nitric anhydride), N₂O₅, does not exist in the free state. It is usually prepared by distilling a mixture of...



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